## $3\alpha$ -(4'-Substituted phenyl)tropane-2 $\beta$ -carboxylic Acid Methyl Esters: Novel Ligands with High Affinity and Selectivity at the Dopamine Transporter

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Since their initial preparation by Clarke and coworkers<sup>1</sup> over 23 years ago, the  $3\beta$ -(substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl ester class of compounds 1 have been widely employed in structureactivity relationship (SAR) studies at the cocaine binding site on the dopamine transporter (DAT).<sup>2</sup> Since neither Clarke's nor any other reported method provided the  $3\alpha$ -phenyl analogs, the SAR studies did not include this isomer. Recently, we reported the synthesis of the first  $3\alpha$ -(substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl esters 2 by samarium iodide reduction of 3-aryl-2carbomethoxytropenes 3.<sup>3</sup> In this paper we describe a more efficient synthesis of several new  $3\alpha$ -(substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl esters **2** and present the results of the first monoamine transporter binding studies on this series of compounds.



The route used to synthesize the  $3\alpha$ -(substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl ester (2a-e) analogs is shown in Scheme 1, and the physical properties are listed in Table 1. Addition of a solution of (1R,5.S)-2-(3'-methyl-1',2',4'-oxadiazol-5'-yl)-8-methyl-8-azabicyclo[3.2.1]oct-2-ene (4)<sup>4</sup> in anhydrous tetrahydrofuran or ether to a -78 °C solution of the appropriate aryllithium followed by quenching with trifluoroacetic acid at -78 °C formed  $3\alpha$ -(substituted phenyl)tropane- $2\alpha$ -(3'-methyl-1',2',4'-oxadiazol-5'-yl)tropanes 5. In some cases ( $X = CH_3$ , I) the  $2\alpha$ , $3\beta$ -isomer was also formed; this isomer could either be removed by flash column chromatography or carried through the next reaction.

Conversion of the oxadiazole to the methyl ester was accomplished by reduction with nickel boride and hydrochloric acid in refluxing methanol. Under the

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Scheme 1



reaction conditions, complete epimerization at C-2 to form the  $2\beta$ , $3\alpha$ -stereoisomer is observed. This is in agreement with previously reported observations on  $3\alpha$ phenyl- $2\alpha$ -(3'-methyl-1',2',4'-oxadiazol-5'-yl)tropane (**5e**, X = H)<sup>5</sup> and is presumably driven by the ability of the piperidine ring to adopt an equatorially substituted twist-boat conformation (Figure 1).

Analysis of the <sup>1</sup>H NMR and COSY spectra of methyl esters **2a**-**e** (Table 2), using pyridine- $d_5$  for chemical shift dispersion, shows weak coupling between H<sub>1</sub> and H<sub>2</sub> ( $J_{1,2} = \sim 1.7$  Hz), while H<sub>2</sub> couples to H<sub>3</sub> with a coupling constant of 9.4–9.7 Hz. These observations are consistent with a twist-boat conformation in which H<sub>2</sub> is close to orthogonal with H<sub>1</sub> and has a near transdiaxial relationship to H<sub>3</sub>. This contrasts sharply with allococaine (**6**) where the <sup>1</sup>H and <sup>13</sup>C NMR data show that this compound possesses a chair conformation.<sup>6</sup> As in the  $2\beta$ , $3\beta$ -isomers **1**, the equatorial proton at C-4 of the  $2\beta$ , $3\alpha$ -isomers **2** is deshielded relative to its axial



## Figure 1.

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Table 1. Chemical Yields and Physical Properties for Oxadiazoles 5 and Methyl Esters 2

synthesis	yield	methyl	yield	molecular	melting	[α] <sub>D</sub> (deg)
oxadiazole	(%)	ester	(%)	formula <sup>a,b</sup>	point <sup>a</sup> (°C)	(c, CH <sub>3</sub> OH)
5a 5b 5c 5d 5e	$egin{array}{c} 85 \\ 23^d \\ 81 \\ 65^e \\ 86 \end{array}$	2a 2b 2c 2d 2e	86 45 54 51 60	$\begin{array}{c} C_{20}H_{26}FNO_8{}{}\cdot}0.5H_2O^c\\ C_{23}H_{28}INO_5S\\ C_{23}H_{28}CINO_5S\\ C_{24}H_{31}NO_5S\\ C_{16}H_{22}CINO_2^f \end{array}$	65 dec 204.6–205.3 182.4–183.8 174.8–175.4 178–179	$\begin{array}{c} -34.4 \ (0.54) \\ -38.9 \ (1.30) \\ -39.9 \ (0.45) \\ -43.1 \ (0.55) \\ -48.1 \ (1.38) \end{array}$

<sup>*a*</sup> Compounds were characterized as the tosylate salts unless otherwise noted. <sup>*b*</sup> C, H, and N analyses were within 0.4% of their theoretical values. <sup>*c*</sup> Characterized as the tartrate salt. <sup>*d*</sup> The  $2\alpha$ ,  $3\beta$ -isomer was recovered in 20% yield. <sup>*e*</sup> The  $2\alpha$ ,  $3\beta$ -isomer was recovered in 15% yield. <sup>*f*</sup> Characterized as the HCl salt.

 Table 2.
 <sup>1</sup>H NMR Selected Chemical Shifts and Coupling Constants for the Free Bases of Compounds 2a-d



			chemical shifts (ppm) <sup>a</sup>					coupling constants (Hz)		
compd	Х	1	2	3	$4_{eq}$	$4_{ax}$	5	$J_{1,2}$	$J_{2,3}$	$J_{3,4_{ m eq}}$
2a	F	3.41	2.65	3.63	2.37	1.32	3.09	1.8	9.7	7.9
2b	Ι	3.40	2.66	3.57	2.35	1.29	3.07	1.7	9.4	7.9
2c	Cl	3.41	2.65	3.59	2.35	1.30	3.08	1.7	9.4	8.0
2d	$CH_3$	3.42	2.72	3.65	2.40	1.35	3.09	1.7	9.3	8.1
2e	Н	3.43	2.73	3.68	2.41	1.40	3.10	1.6	9.5	8.1

 $^a$  Chemical shifts are reported relative to Si(CH\_3)\_4 in pyridine-  $d_5$  at 500 MHz.

counterpart, indicating that the aromatic ring lies perpendicular to the axis of the piperidine ring.

The IC<sub>50</sub> values for the inhibition of ligand binding to the dopamine, serotonin, and norepinephrine transporters by  $3\alpha$ -(*p*-substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl esters **2a**-**e** are listed in Table 3. For comparison, the previously reported IC<sub>50</sub> values for the corresponding  $2\beta$ ,  $3\beta$ -isomers **1a**-**e**, as well as values for cocaine (**7**) and allococaine (**6**), are also listed. The IC<sub>50</sub> values for dopamine and serotonin represent inhibition of 0.5 nM [<sup>3</sup>H]WIN 35,428 and 0.2 nM [<sup>3</sup>H]paroxetine, respectively, as previously described.<sup>7</sup> Norepinephrine IC<sub>50</sub> values represent inhibition of 0.5 nM [<sup>3</sup>H]nisoxetine binding to the norepinephrine transporter.<sup>8,9</sup>

The substituted aryl  $2\beta$ , $3\alpha$ -isomers **2a**-**d** are more selective for the DAT relative to the 5-HT transporter than the  $2\beta$ , $3\beta$ -isomers **1a**-**d**. The unsubstituted analog **2e** possessed DAT selectivity similar to WIN 35,065-



2. Furthermore, the  $2\beta$ ,  $3\alpha$ -isomers are only 1.5–5.9 times less potent at the DAT than the analogous  $2\beta$ ,  $3\beta$ isomers. These results contrast sharply with allococaine (6), the  $2\beta$ ,  $3\alpha$ -stereoisomer of cocaine (7), which is 60 times less potent than cocaine at the DAT.<sup>10</sup> Apparently, the preference of  $2\beta$ ,  $3\alpha$ -esters **2** for the twist-boat conformation allows the amino, aryl, and carbomethoxy groups to adopt similar positions to the corresponding groups in 1 and may explain the relatively high dopamine binding affinities exhibited by esters 2 relative to allococaine (6). It is interesting to note that the piodophenyl analog 2b (RTI-352) is only slightly less potent than 1b (RTI-55) but 7 times more selective for the DAT relative to the 5-HT transporter. Since [123I]-RTI-55 has proven to be a valuable SPECT (singlephoton emission-computed tomography) imaging agent,<sup>11</sup> we plan to conduct further studies with 2b to determine if it possesses properties that would make it a better SPECT imaging agent than [123]RTI-55.

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**Supporting Information Available:** Experimental data for **5a**–**d** and **2a**–**d** (7 pages). Ordering information is found on any current masthead page.

Table 3. Comparison of Transporter Binding Potencies for Stereoisomers of WIN 35,065-2 Analogs

-		-	-					
			$IC_{50} (nM)^{a}$					
			DA NE 5-HT			ratio <sup>b</sup>		
compd	Х	isomer	[ <sup>3</sup> H]WIN 35,428	[ <sup>3</sup> H]nisoxetine	[ <sup>3</sup> H]paroxetine	NE/DA	5-HT/DA	
6 (allococaine) <sup>c</sup>			$6160\pm900$					
7 (cocaine) $^d$			$89.1 \pm 4.8$	$3298 \pm 293$	$1045\pm89$			
1a <sup>e,f</sup>	F	$2\beta, 3\beta$	$14\pm1.4$	$835\pm45$	$810\pm59$	60	58	
2a	F	$2\beta,3\alpha$	$21\pm0.57$	$1230\pm91$	$5060 \pm 485$	59	241	
1 <b>b</b> <sup>e</sup>	Ι	$2\beta, 3\beta$	$1.26\pm0.04$	$36\pm2.7$	$4.21\pm0.3$	29	3.3	
2b	Ι	$2\beta, 3\alpha$	$2.85\pm0.16$	$52.4\pm4.9$	$64.9 \pm 1.97$	18	23	
$\mathbf{1c}^{e}$	Cl	$2\beta, 3\beta$	$1.12\pm0.1$	$37\pm2.1$	$44.5\pm1.3$	33	40	
2c	Cl	$2\beta, 3\alpha$	$2.4\pm0.2$	$60.1 \pm 2.4$	$998 \pm 120$	25	415	
$1d^e$	$CH_3$	$2\beta, 3\beta$	$1.71\pm0.31$	$60\pm0.53$	$240\pm27$	35	140	
2d	$CH_3$	$2\beta,3\alpha$	$10.2\pm0.8$	$275\pm24$	$4250\pm422$	27	417	
1e <sup>g,h</sup>	Н	$2\beta, 3\beta$	$23\pm5$	$920\pm73$	$2000\pm64$	40	87	
2e	Н	$2\beta,3\alpha$	$101\pm16$	$2080 \pm 285$	$5700\pm721$	21	57	

<sup>*a*</sup> Data are mean  $\pm$  standard error of three or four experiments performed in triplicate. <sup>*b*</sup> Ratios of IC<sub>50</sub> values. <sup>*c*</sup> IC<sub>50</sub> values taken from ref 10. <sup>*d*</sup> IC<sub>50</sub> values taken from ref 7. <sup>*e*</sup> IC<sub>50</sub> values taken from ref 12. <sup>*f*</sup> This compound is WIN 35,428. <sup>*g*</sup> This compound is WIN 35,065-2. <sup>*h*</sup> IC<sub>50</sub> values taken from ref 13.

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